

- Opening
  - Introduction
  - Topic: Quantum Mechanics
- Base Significance
  - Going from large scale to small
  - Previous topics, galaxies, black holes, universes
  - Quantum Mechanics is the smallest of scales
  - Not the domain of classical physics.
- Introduction
  - Entirely new way of analyzing the natural world.
  - Required a new generation of brave new intellectuals to tackle old problems.
  - It is not a subject of exact values, constants, simple kinematic mechanics, it is one of probability, waves, and uncertainty.
  - Early quantum theory led to computer breakthroughs, LED's, lasers, medical imaging, and superconducting magnets necessary for MagLev trains.
- Early formation
  - The modern concept was invented in the early 20th century by Erwin Schrodinger, Max Planck, Werner Heisenberg, Max Born, and other imaginative scientists studying the mysteries of black body radiation.
    - They were also driven by the discovery that subatomic particles do not act like they ought, almost like they were waves.
  - Einstein's 1905 paper on the photoelectric effect employed the precursors to quantum theory to explain the phenomenon
    - Relevant later in Einstein's career
    - Photoelectric effect: Emission of electrons or other loosely bonded subatomic particles when light strikes a material.
  - Early physicists did not realize what they were creating
    - Max Planck considered his famous equation relating energy elements and frequency of a wave to simply be a mathematical trick, and not related to reality.
    - Einstein took this and managed to show that EM waves can be described as a waves or particles with discrete quanta of energy which is dependant on frequency.
    - This was highly debated among the scientific community, at the end it was accepted along with much of the basic quantum mechanics which were still considered to be fringe physics.
    - Later refined to show that subatomic particles and EM waves are very similar in their neither are ever wholly particle or wholly wave
- Applications in Science
  - Bohr's Atom
    - Electron orbitals are not sure things, they are guesses
    - Orbitals are man made things based on discrete energy levels.

- They are not clear cut orbits like those of planets, they are small bands of probability that an electron will or will not be on them.
  - Heisenberg's Uncertainty Principle
    - Derivation of the wave-particle duality of EM waves and subatomic particles
    - States that in an experiment, only one of the two states can be tested, therefore when observed, a particle or EM field is forced into a wave or a particle, therefore it is impossible to know all of the values of all of the properties of a system at the same moment.
    - This leads to making the field a study a probabilities, if they can not be known, they must be guessed.
    - Heisenberg surmises that certain physical properties are shared, such as momentum and position, and both cannot be known at the same moment, that is, they cannot be simultaneously measured.
  - Electron Spin
    - The result of Wolfgang Pauli and his exclusion principle
    - Pauli's solution to resolve the inconsistencies between molecular behavior and quantum mechanics
    - Essentially added another value to electrons, another discrete level of information
  - Quantum Entanglement
    - The coolest thing since the invention of podcasts
    - The Holy Grail of space nerds in quantum mechanics
    - Two particles, such as photons, interact physically.
    - They form a connection that is not well understood, but essentially means that one will mimic the other's state, spin, etc. and the other way around.
    - Scientists have measured the propagation of this phenomenon at speeds greater than 10,000 times the speed of light.
      - The transfer is theoretically instantaneous over any distance, but that is only a theory for now.
    - Essentially flaunts the universal speed limit.
    - There is now a caveat to that law, nothing can move faster than light, except information.
    - Infamously name "spooky action at a distance"
      - Marks Einstein's derision for quantum mechanics later in life.
- Summary
  - If force and energy are the bases for Newtonian Physics, information is the foundation of quantum mechanics
  - Information cannot be created or destroyed, only transferred.
    - Led to the discovery of Hawking Radiation